

Membrane Technology

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Chapter 1

Introduction to Membrane Technology



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Chapter Description

- Aims
 - Understand what is membrane, terminology and membrane types.
- Expected Outcomes
 - Understand membrane in general.
- Other related Information



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Subtopics

- 1.1 Membrane Definition**
- 1.2 Membrane Terminology**
- 1.3 Membrane Selectivity**
- 1.4 Driving Force**
- 1.5 Membrane Types**



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1.1 Membrane Definition

- Membrane – the original word: “**membrana** “ (Latin, which mean skin)
- Other definition of membrane:
 - A barrier which prevent mass transport but allows restricted and / or regulated passage of one or more species.



1.2 Membrane Terminology

- Feed - the solution to be separated or processed
- Permeate - the filtrate, the liquid or solid or gas that passing through the membrane
- Retentate - the concentrate, the retained solute



1.2 Membrane Terminology

- Flux – the rate of volume or mass or mole permeate per effective membrane area per time (i.e. L/m².h)
- Permeability- transport flux per unit transmembrane pressure (i.e. L/m².h.bar)
- Rejection- percent of solute removal
- Separation Factor- composition ratio of the permeate components (i and j) relative to its composition ratio in the retentate side



1.3 Membrane Selectivity

- The membrane's selectivity affected by:
 - Solute Shape
 - Solute Size
 - Electrostatic charge effect
 - Solute Diffusivity
 - Physicochemical interactions
 - Volatility
 - Polarity/solubility



1.4 Driving Force

1. Most of the membrane technology is based on Transmembrane pressure (TMP)
2. Concentration gradient- membrane for hemodialysis
3. Other driving forces are including:
 - Chemical potential
 - Osmotic pressure
 - Electric field
 - Magnetic field
 - Partial pressure
 - pH gradient



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1.5 Type of Membranes

	Cut-offs of different liquid filtration techniques									
Micrometer logarithmic scaled	0,001	0,01	0,1	1	10	100	1000			
Angstroms logarithmic scaled	1	10	100	1000	10 ⁴	10 ⁵	10 ⁶	10 ⁷		
Molecular weight (Dextran in kD)	0,5	50	7.000							
Size ratio of substances to be separated	Solved salts	Sugar	Pyrogens	Viruses	Albumin (66 kD)	Bacteria	Yeast	Pollen	Human hair	Sand
Separating process	Reverse osmosis		Nano filtration	Ultra filtration		Micro filtration			Particle filtration	

https://upload.wikimedia.org/wikipedia/commons/e/e7/Cut-offs_of_different_liquid_filtration_techniques.png



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