

## Oleochemistry

#### Surfactant

by Shamsul Bin Zakaria Faculty Industrial Science and Technology shamsulzakaria@ump.edu.my



# Student should be able to understand and compare:

- The classes of surfactants
- The main example of every classes of surfactants
- The applications of the surfactants



#### Definition....

• Surfactants are compounds that lower the surface tension of a liquid and lowering of the interfacial tension between two liquids, or between a liquid and a solid. Surfactants may act as detergents, wetting agents, emulsifiers, foaming agents, and dispersants.



## Surfactant classes

- ANIONICS
- AMPHOTERICS/BETAINES
- CATIONICS
- NONIONICS



# Anionics

Definition: The anion portion of the molecule is lipophilic.

### Example: Sodium Lauryl Sulfate: Cation: Sodium (hydrophilic)

Anion: LauryIsulfate (lipophilic)



# Most commonly used anionics

- 1. Alcohol Sulfates.
- 2. Ethoxylated Alcohol Sulfates.
- 3. Sulfosuccinates.
- 4. Linear Alkylbenzene Sulfonates.
- 5. Alpha Olefin Sulfonates.



#### Alcohol sulfates

- History: Commercial usage began in the mid 1940's.
- Made by the reaction of a fatty alcohol (C8-C18) with sulfur trioxide.
- The alcohol sulfuric acid is neutralized with sodium or ammonium hydroxide, or an amine.



# Characteristics of alcohol sulfates

- 1. C12 (Lauryl) provides the highest foam.
- 2. Hydrolyzes at pH < 4.5.
- 3. Can be readily thickened when combined with alkanolamides, betaines, amine oxides etc.
- 4. Thickening can be greatly enhanced by the addition of electrolytes
- 5. (NaCl, KCl etc.).
- 6. Light color.
- 7. Bland odor.



# **Major applications**

- Shampoos
- Body Wash
- Liquid Hand Soap
- Bath Products
- Facial Cleansers
- Syndet Bars



# Properties of amphotericsfor hi&i applications

- Stable in alkaline and acid conditions.
- The propionate and dipropionate type are excellent for HI&I cleaners.
- Low and high foaming is dependent on the molecular weight of the fatty moiety.
- A C8 will be a low foamer, whereas, a C12 is a high foamer.



## Examples of caustic solubility

<u>Compound</u>	<u>%NaOH</u>
Coco Dipropionate	41
Caprylic (C8) Dipropionate	38
Caprylic (C8) Propionate	30
Caprylic (C8) Diacetate	28
Coco Propionate	25
Octyl Betaine	23
Sodium Xylene Sulfonate	22



## Comparative foam heights

<u>Compound</u>	<u>Foam Ht.</u>
Cocamidopropyl Betaine	210
Coco Propionate	155
Coco Dipropionate	145
Octyl (C8) Betaine	50
Capryl (C8) amidopropyl Betaine	30
Caprylic (C8) Diacetate	10



# Properties of amphoterics for personal cleansing

- Stable over a wide pH range.
- Mild to skin and eye.
- Reduce irritation of ether sulfates.



#### Cationic surfactants

Definition: The cation is the lipophilic portion of the molecule.

Types:

- Quaternary Ammonium Compounds
- Amine Salts



# Properties of quaternary ammonium compounds

- Lower molecular weight are typically used as biocides.
- Higher molecular weight (C18) are excellent hair conditioners.
- Most are incompatible with anionic surfactants.
- Low foaming.
- Extremely sensitive to hard water and usually require a chelant.



# Major uses of "quats"

- Biocides.
- Fabric Softeners.
- Hair Conditioners.
- Antistatic Agents.
- "Cheater" Wax.
- Corrosion Inhibitors.
- Leather Softening.
- Pigment Dispersants.
- Sewage Flocculants.





# The most commonly used for household and industrial applications:

# Lauryl dimethyl benzyl ammonium chloride



# Biocidal "quats"

#### MODE OF ACTION

- 1. Reduce surface tension at interface.
- 2. Attracted to negatively charged surfaces, including microorganisms.
- 3. Denature protein of bacterial or fungi cells.
- 4. Affect the metabolic reactions of the cell.
- 5. Vital substances leak out.
- 6. Causes death.



## Fabric softeners

## Most widely used

• Distearyl dimethyl ammonium chloride.

Dialkyl imidazolinium methyl methoslfate.



#### MOST WIDELY USED "QUATS" USED IN HAIR CONDITIONERS

- Cetrimonium Chloride
- Stearalkonium Chloride
- Distearyldimonium Chloride



# Nonionics

- Alkanol Amides.
- Amine Oxides.
- Ethoxylated Nonyl Phenol or Alcohols.



#### Preparation of alkanol amides

Made by the reaction of a mono or diethanol amine with a fatty acid, methylester or fatty glyceride, (e.g., coconut oil).



# Alkanol amides

- Most cost/effective thickener and foam stablizer available.
- History: Commercially available in the mid 1940's.
- Diethanolamides are being phased out of formulas due to reported "cancer link".
- They are being replaced by: Betaines, Amine Oxides, Monoethanolamides and Monoisopropanolamides.



# Some new guys on the block "natural surfactants"

- Decylglucoside: Derived from sugar and coconut oil.
- Cocoyl Glutamate: Derived from glutamic acid (amino acid) and coconut oil.
- Cocosulfate: Derived from coconut oil.



## **Current trends and limitations**

- Natural.
- Certified Organic.
- Animal friendly.
- DEA Free.
- Formaldehyde Free.
- Nitrosamine Free.
- Sulfate Free.
- Low Dioxane.



#### Conclusion

- Surfactants have four classes
- Every classes have particular properties
- Natural surfactant is the emerging research in surface chemistry





## **Chapter description**

All pictures/photographs/diagrams/figures used in this chapter is subjected to common creative that for education purposes

