

### **Air Pollution Control Technology**

## Air Pollution control : Gases and Vapors

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#### Introduction

- Aim
  - This chapter discuss a methods used to control air pollution from gas stream
- Outcome
  - Students are able to discuss common methods for air pollution control in gases stream







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- Units of measure:
  - 1) Micrograms per cubic meter (µg/m<sup>3</sup>)
  - 2) Parts per million (ppm)
  - 3) Micron ( $\mu$ ) / micrometer ( $\mu$ m)
- ppm in water and wastewater, mass-to-mass ratio
- ppm in air, volume-to-volume ratio

ppm = [ ( $M_p$ /GMW) x 22.414 x ( $T_2$ /273K) x (101.325 kPa /  $P_2$  ) ] / [  $V_a$  x 1000 L/m<sup>3</sup>) ]





• Example :

A one-cubic-metre sample of air was found to contain 80  $\mu$ g/m<sup>3</sup> of SO<sub>2</sub>. The temperature and pressure were 25°C and 103.193 kPa when the air sample was taken. What was the SO<sub>2</sub> concentration in ppm

GMW of  $SO_2 = 32.06 + 2 (15.99994) = 64.06$ Answer : 0.0300 ppm of  $SO_2$ 



### Regulation

- Performance monitoring
- Maintenance record
- Opacity
- Limit values and technical standard
- Hazardous substance
- Periodic monitoring
- CEM
- Emission declaration





- Techniques without using emissions control devices
  - Process change
  - Wind, geothermal, hydroelectric or solar unit
  - Change in fuel
  - Good operating practices
    - Good house keeping
    - Maintanance
  - Plant shutdown



#### Gases pollution and control







- Flue gas desulfurizing process
  - Limestone scrubbing
  - Lime scrubbing
  - Dual alkali processes
  - Lie spray dying
  - Wellman-lord process





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