## General Chemistry

## Matter

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## Question 1

- Calculate molar concentration of a 1.5 L solution containing 4.00 g $\mathrm{NaOH} .(\mathrm{NaOH}$ molecular weight $=40)$


## Question 2

- Calculate molarity concentration of H 2 SO 4 solution with molality of 8.010 and a density of $1.354 \mathrm{~g} / \mathrm{mL}$.

[^0]
[^0]:    A solution of $\mathrm{H}_{2} \mathrm{SO}_{4}$ with a molal concentration of 8.010 m has a density of $1.354 \mathrm{~g} / \mathrm{mL}$. What is the molar concentration of this solution?

