

# BFF1113

# Engineering Materials



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# ENGINEERING MATERIALS FOR MANUFACTURING ENGINEERS

# *Reference books:*

William D. Callister, Jr.  
Materials Science & Engineering: An Introduction.  
John Wiley & Sons Inc.

Serope Kalpakjian, Steven R. Schmid.  
Manufacturing Engineering and Technology  
Pearson Prentice-Hall, Inc.

1	Course Code and Name	BFF1113 ENGINEERING MATERIALS		
2	Semester / Year	Semester 1 / Year 1		
3	Program Level / Category	Bachelor Degree / Materials		
4	Unit	3 Credits		
5	Prerequisite Course	NIL		
6	Contact Hours	Lecture	2 unit	2 H X 14 weeks
		Tutorial	0.5 unit	1 H X 14 weeks
		Laboratory	0.5 unit	1 H X 14 weeks
7	Course Synopsis	<p>This course introduces the fundamental concepts of engineering materials which includes the structure of materials, mechanical and physical properties of materials, binary phase diagrams, TTT diagram, heat treatment, applications and current developments of metal, polymer, ceramic, composite and advanced materials. Also, basic understanding on the environmental degradation of engineering materials.</p>		
8	Course Outcomes	<p>By the end of semester, students should be able to:</p> <p>CO1: Gain fundamental concepts on the classification of various engineering materials.</p> <p>CO2: Perform structure-property correlations for engineering materials based on the microstructures.</p> <p>CO3: Analyse binary phase diagrams, TTT diagram, cooling curve, and correlate with microstructures and properties.</p> <p>CO4: Evaluate the criteria in the selection of materials.</p> <p>CO5: Recognize the importance of sustainable materials.</p>		

9	CO-PO Mapping	CO/PO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	
		CO1	✓												
		CO2		✓											
		CO3		✓											
		CO4			✓										
		CO5							✓						
10	Assessment Methods	Distribution (%)				CO1		CO2		CO3		CO4		CO5	
		Assignment				15%		✓		✓		✓		✓	
		Test 1				15%		✓		✓					
		Test 2				15%				✓		✓			
		Final Examination				40%		✓		✓		✓		✓	
		Lab Report				15%		✓		✓					
		Total				100%									

PO	1	2	3	4	5	6	7	8	9	10	11	12
	Knowledge	Problem Analysis	Design/Development of Solution	Investigation	Modern Tool Usage	Engineer & Society	Environment and Sustainability	Ethics	Communication	Individual & Teamwork	Life Long Learning	Project Management and Finance

# Contribution of assessment

	DISTRIBUTION (%)	REMARKS
1) ASSIGNMENTS	15 %	<ul style="list-style-type: none"><li>• +- 3 assignments</li><li>• submit on time. Late submission not ALLOWED.</li><li>• student to solve each assignment <b>independently. No copy and paste.</b> Discussion are allowed.</li></ul>
2) LAB ASSIGNMENTS	15 %	<ul style="list-style-type: none"><li>• +- 2 lab assignments</li><li>• perform experiment in the materials lab.</li></ul>
3) TEST 1	30 %	<ul style="list-style-type: none"><li>• given during semester after a sizeable of lessons are learned.</li><li>• 1 hour and 30 minutes test.</li></ul>
4) TEST 2		
5) FINAL EXAMINATION	40 %	<ul style="list-style-type: none"><li>• 3 hours examination with 4 questions.</li><li>• contributes 40% of the final grade.</li></ul>

# Course Guidelines:

1. Introduction to Engineering Materials
2. Bonding and Properties
3. Crystal Structures & Properties
4. Imperfection in Solids
5. Mechanical Properties of Materials
6. Physical Properties of Materials
7. Failure & Fundamental of Fracture
8. Metal Alloys
9. Phase Diagram
10. Phase Transformation – Heat Treatment
11. Processing and Application of Metals
12. Ceramic Materials
13. Polymer Materials
14. Composite Materials
15. Corrosion & Degradation of Materials
16. Environment and Sustainability

# Introduction to Engineering Materials



# What are Materials?

## ■ Materials

- Substances from which something is composed or made from.
- Materials can be natural - like wood, or human made - like plastic
- There are now more than 300,000 different known materials



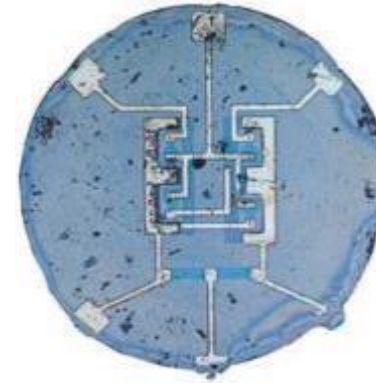
# What are Engineering Materials?

## ■ Engineering Materials

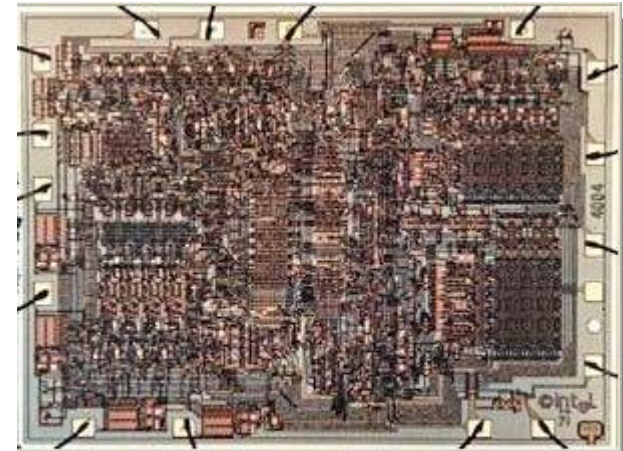
- Materials used to produce technical products.
- Engineered materials with desired properties.

# ■ 1.1 Materials in our daily life

## ◆ Electrical & Electronic



The first commercially produced silicon transistor, developed by Texas Instruments in the early 1950s. Courtesy: Texas Instruments.



The Intel 4004 microprocessor, which was introduced in 1971. The 4004 contained only 2300 transistors and performed 60,000 calculations per second. Courtesy: Intel.

# Civil & Structural



# Aerospace



# Chemical



# Infrastructural & Transportation

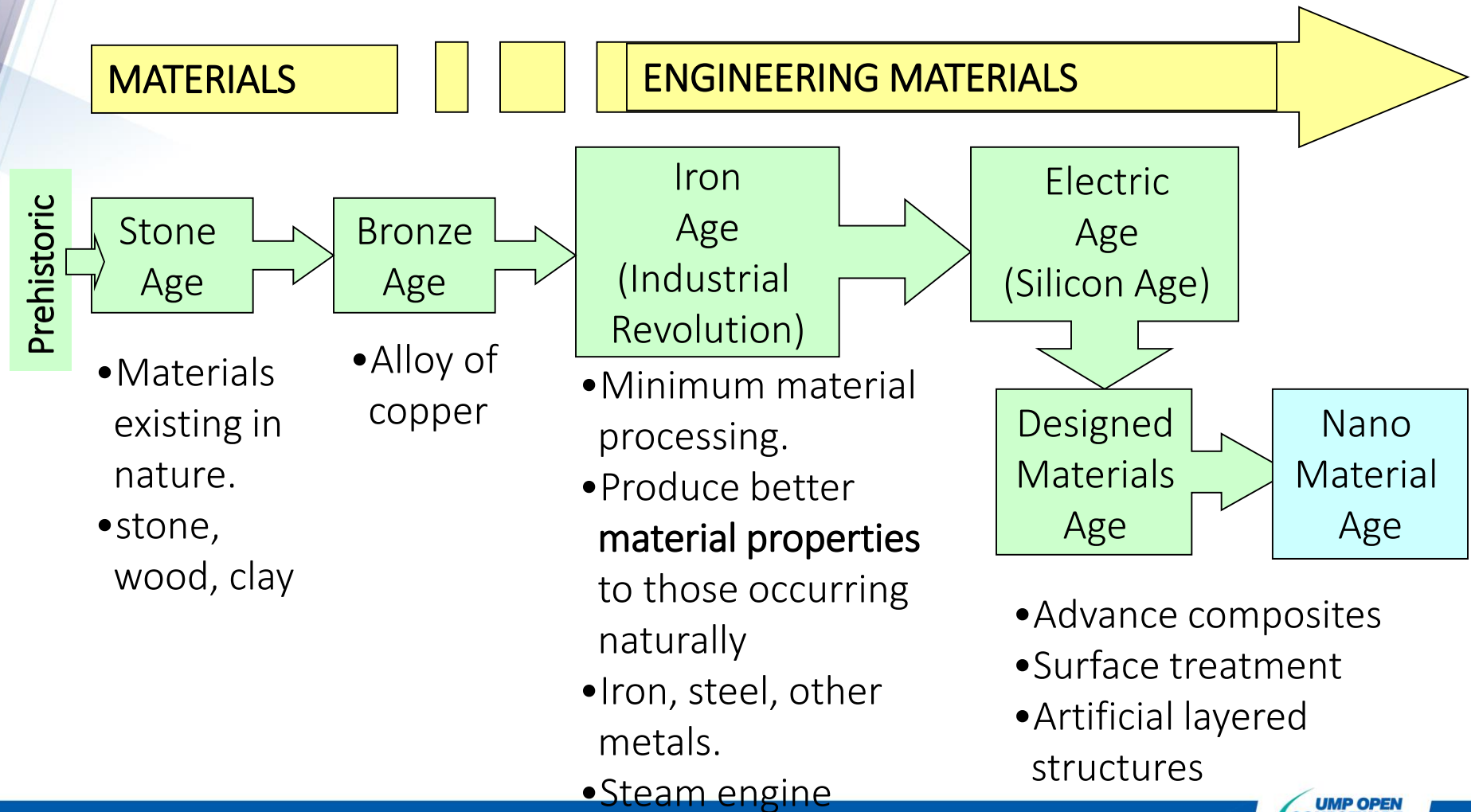






■ 1.2 Historical development  
of materials

# Evolution of Materials



# Prehistoric

Stone, clay, wood



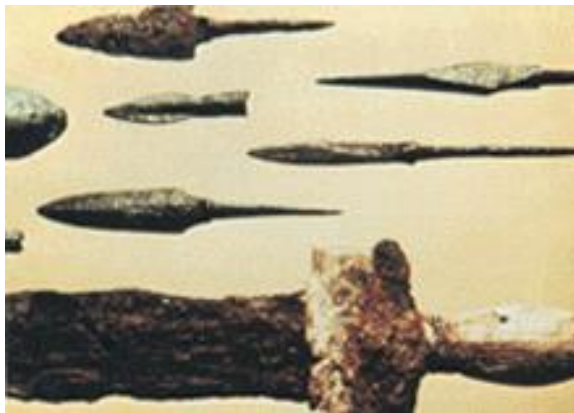
## Stone Age

- Early in the developments of human cultures, before the use of metals
- Tools & weapons were made of stone



## Bronze Age

- Bronze (Copper + tin + zinc)
- The time in the development of any human culture
- Before introduction of iron, when most tools and weapons were made from bronze



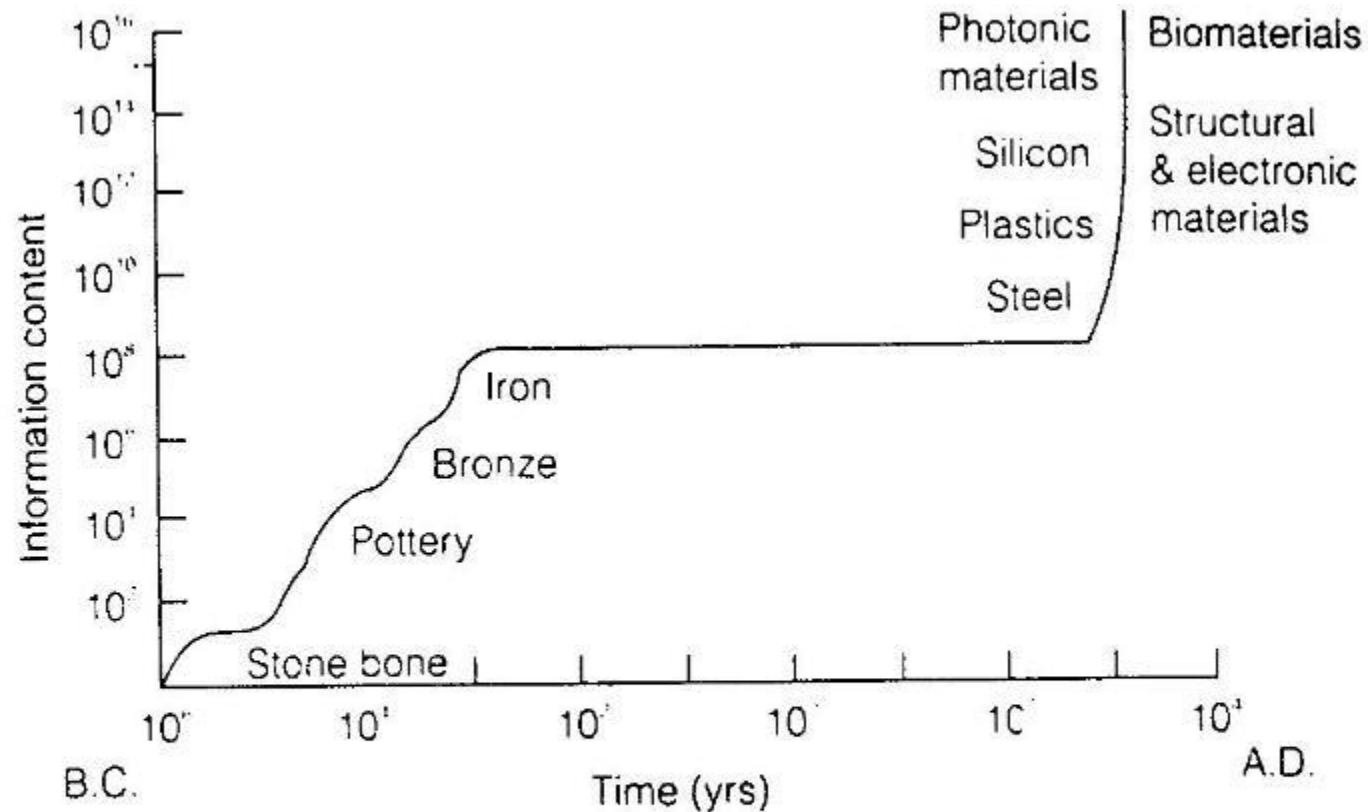
## Iron Age

- Marks the period of development of technology replacing bronze as the basic material for tool implementation and weapon
- Last stage of the archaeological sequence



# Electrical Age (Silicon Age) and beyond





**Fig. 1.4** A schematic diagram showing the growth in technology information content inherent in the use of new materials. It emphasizes the present rapidly changing circumstances before reaching another plateau when the second industrial revolution nears maturity.



# ■ 1.3 Materials science and engineering

# Discipline of Materials Study

## ■ Materials Science

- Science - *scientia* (Knowledge)

- Study of basic materials knowledge.

- Investigation of the relationship between **STRUCTURES** and **PROPERTIES** of materials

## ■ Materials Engineering

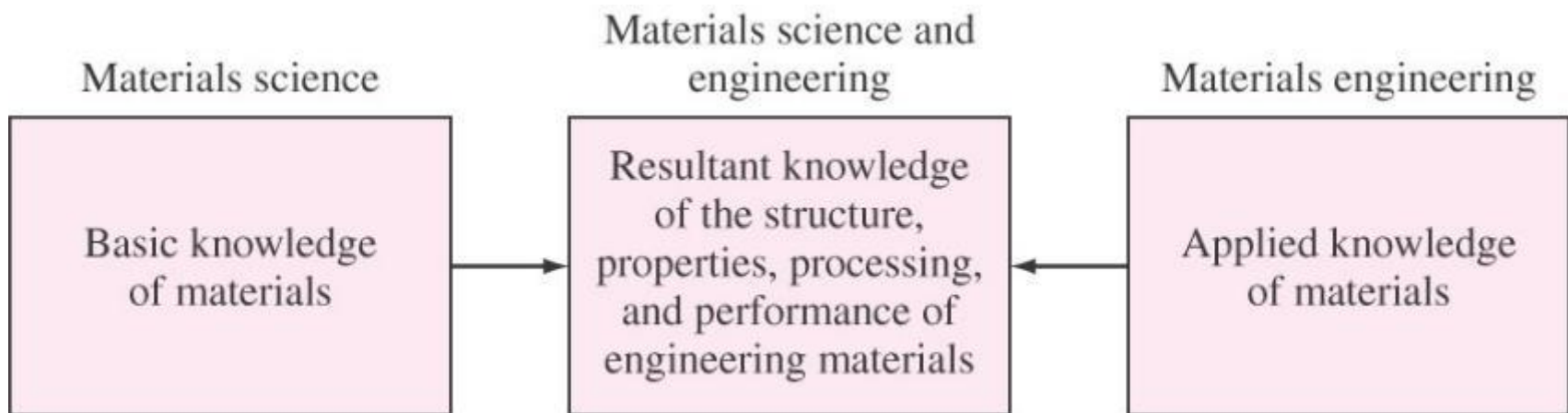
- Engineer – *ingeneur*

- **Used** of Materials Science knowledge (fundamental) to **design** and to **produce** materials with **properties** that will meet the **requirements** of society.

- Structure-Property correlations, designing or engineering the structure of a material to a pre-determined set of properties.

## ■ **Materials Science and Engineering (MSE)**


- Combines both basic knowledge and applications and forms a **bridge** between the basic sciences (physics, chemistry, and mathematic) and the various engineering disciplines (electrical, mechanical, manufacturing, chemical, civil, and aerospace engineering).
- **Interdisciplinary** nature.





# Case Study: Titanic

**The Royal Mail Ship *Titanic*: Did a Metallurgical Failure Cause a Night to Remember?**  
Katherine Felton, H.P. Leighty, Jr., and A. Janovic



**INTRODUCTION**

On the early morning of April 14, 1912, the RMS Titanic departed from Southampton, England, on her maiden voyage to New York City. The ship, which was the largest and most luxurious passenger liner ever built, struck an iceberg in the North Atlantic ocean and sank within two hours and 40 minutes. The ship's sinking resulted in the loss of 1,517 lives, including 710 children and 512 women.

The investigation into the cause of the disaster was one of the most thorough ever conducted. It included a metallurgical analysis of the hull of the Titanic, which revealed that the steel used in the ship's hull was brittle at the time of the collision. The temperature of the sea water was -2°C, and the analysis also showed that the steel used was possibly the best steel available at the time of the ship's construction.



At the time of the collision, the temperature of the sea water was **-2° C**.

Plain carbon steel had a high ductile-brittle transition temperature, making it unsuitable for service at low temperatures;

Video:  
Effect of DBTT to Materials

Yale

# Materials Selection Process

1. Pick **Application** → Determine required **Properties**  
Properties: mechanical, electrical, thermal, magnetic, optical, deteriorative.
2. **Properties** → Identify candidate **Material(s)**  
Material: structure, composition.
3. **Material** → Identify required **Processing**  
Processing: changes *structure* and overall *shape*  
ex: casting, sintering, vapor deposition, doping  
forming, joining, annealing.

## Selecting the RIGHT material.

- To succeed in **processing** materials with a given set of **properties**, it is necessary to understand the basis of the **properties** from the atomic and molecular level, and to understand how small differences in structure can lead to large differences in observed properties.
- 4 components that are involved in design, production, and utilization of materials.
  - **Processing**
  - **Structure**
  - **Properties**
  - **Performance**
- How do these components interrelated?



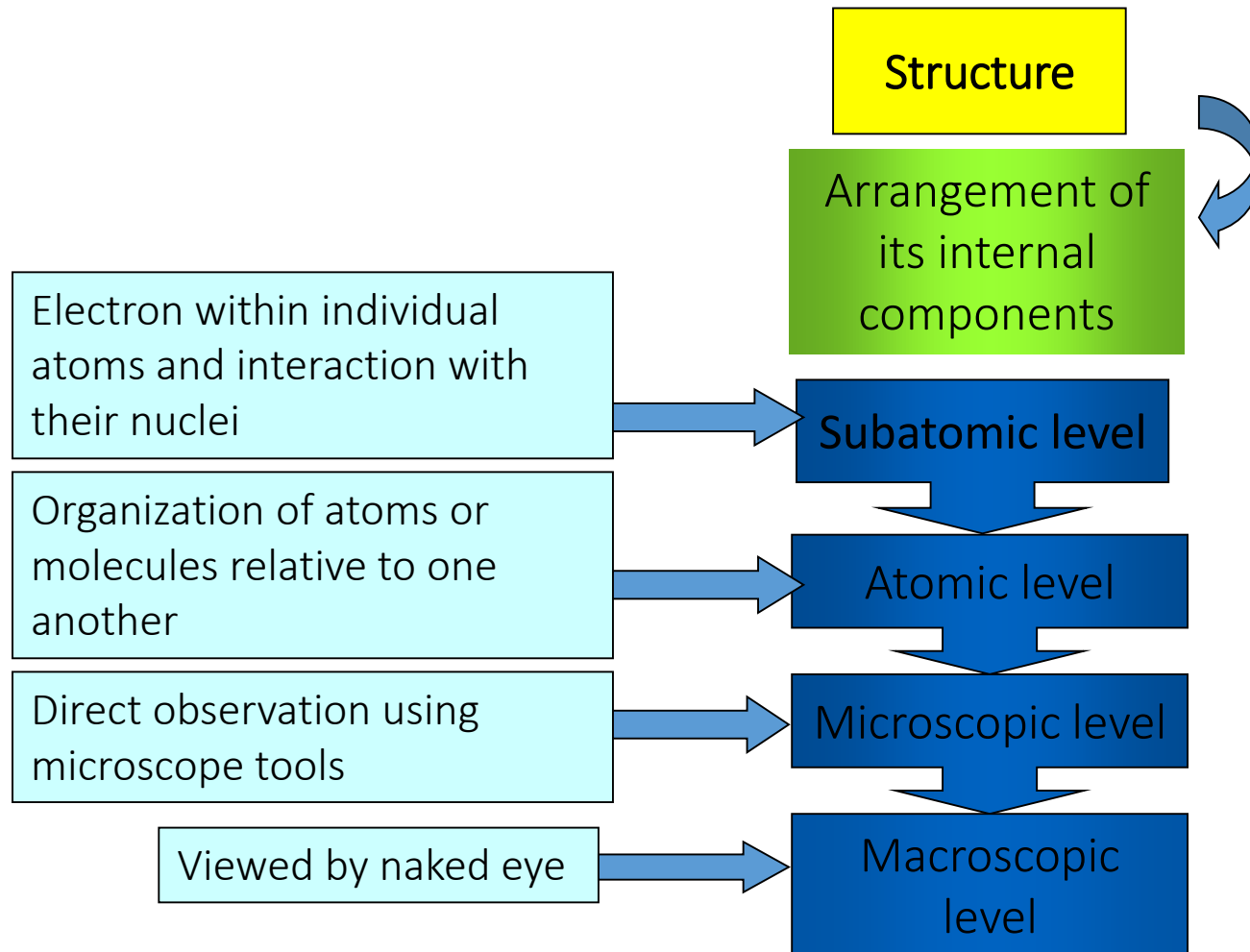
**IMPORTANT**



# 1. What is Processing?

Multiple procedures to produce something pre-determined.

# 2. What is Structure?



# 3.0 What is Property?

## Properties

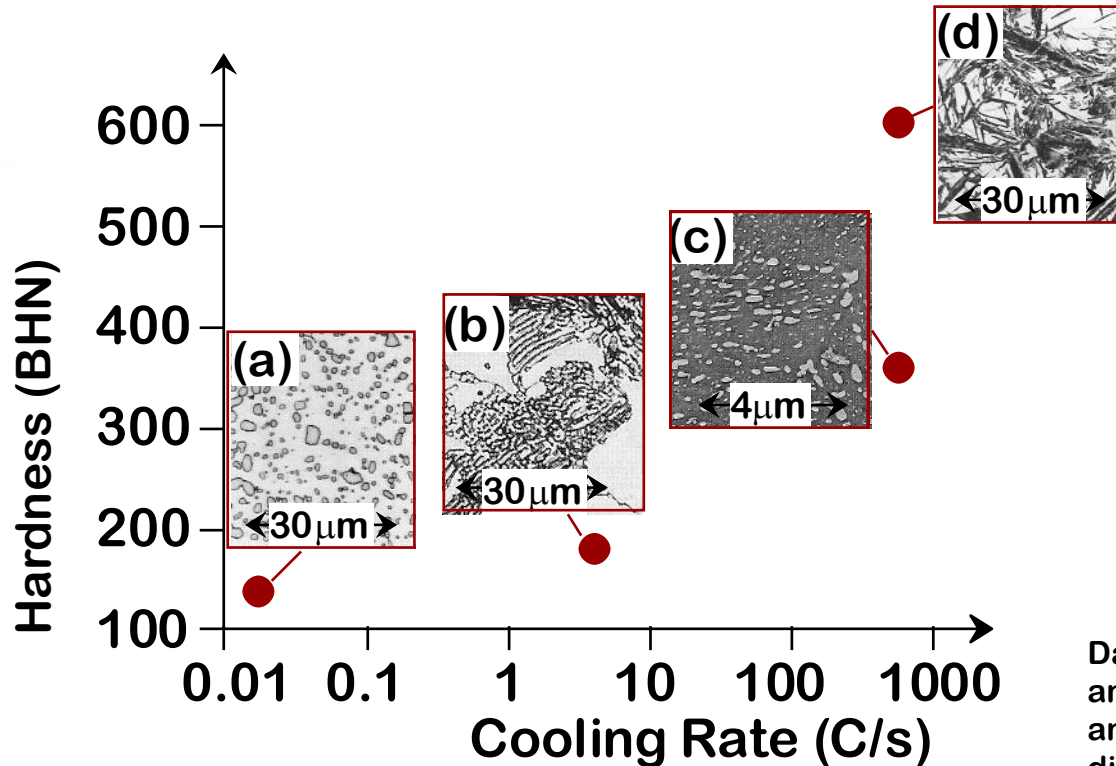
Materials trait in term of the kind and magnitude of response to a specific imposed stimulus

- Material: Conductor
- Stimulus: voltage
- Response: Electric Current

- **PROPERTIES**
  - Mechanical
  - Electrical
  - Thermal
  - Magnetic
  - Optical
  - Deteriorative

Property	Stimulus	Response
Mechanical	Applied load or force	Deformation
Electrical	Electric field	Conductivity

- **Properties** depend on **structure**  
ex: hardness vs structure of steel



Data obtained from Figs. 10.21(a) and 10.23 with 4wt%C composition, and from Fig. 11.13 and associated discussion, *Callister 6e*.

Micrographs adapted from (a) Fig. 10.10; (b) Fig. 9.27; (c) Fig. 10.24; and (d) Fig. 10.12, *Callister 6e*.

- **Processing** can change **structure**  
ex: structure vs cooling rate of steel

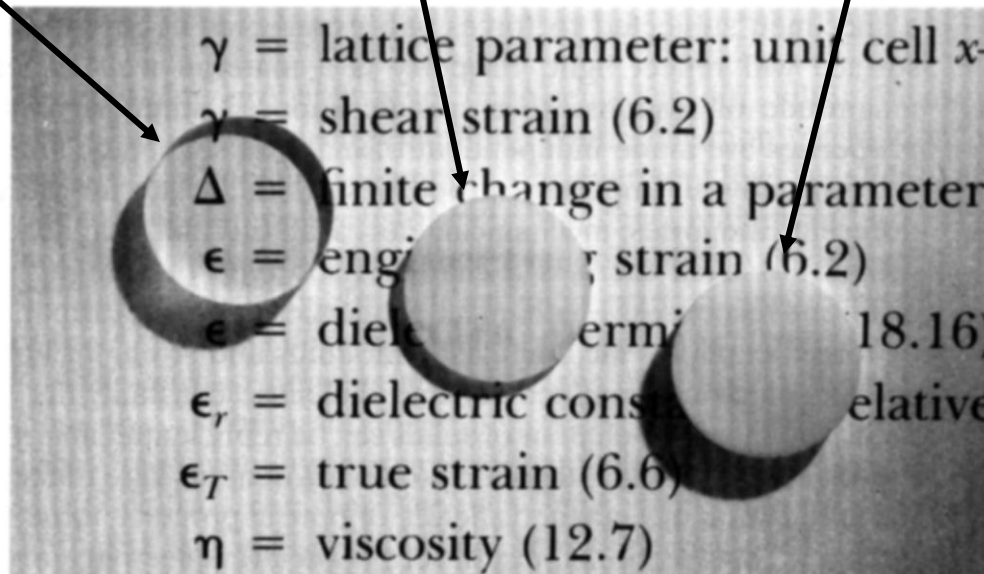
# Optical property

- **Transmittance:**
  - Aluminum oxide may be transparent, translucent, or opaque depending on the material structure.

single crystal

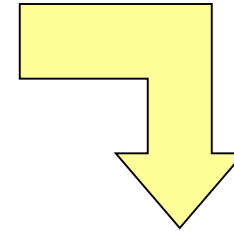
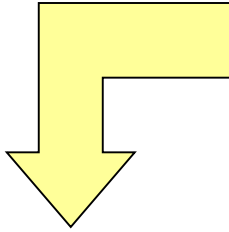
polycrystal:  
low porosity

polycrystal:  
high porosity



Adapted from Fig. 1.2,  
*Callister 6e*.  
(Specimen preparation,  
P.A. Lessing; photo by J.  
Telford.)

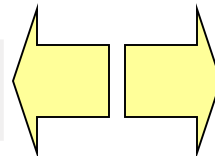
**Change of Properties**



**Heat Treatment**

**Adding other substances**

**Structure**

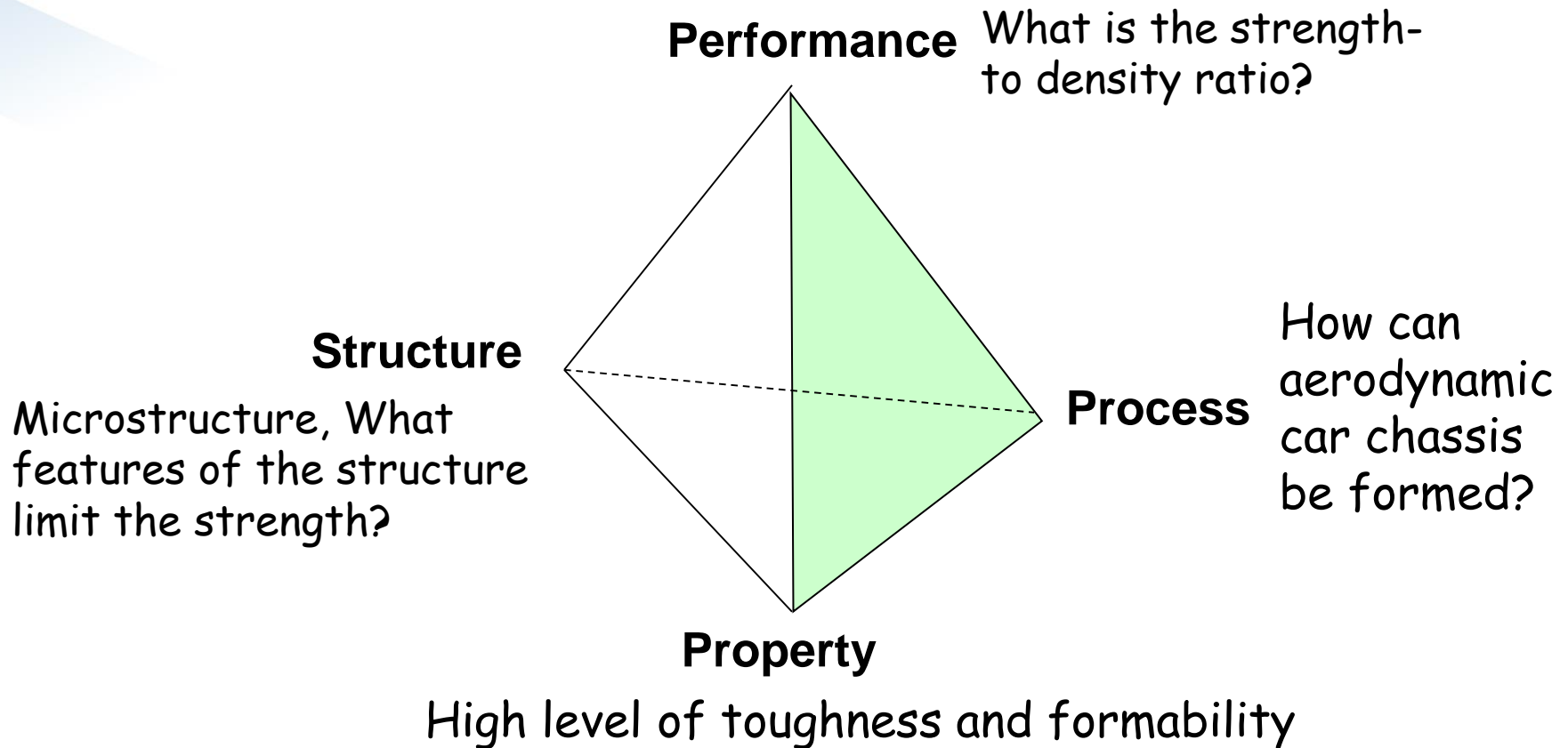


**Properties**

# 4. What is Performance?

A measurement of how good a product is.

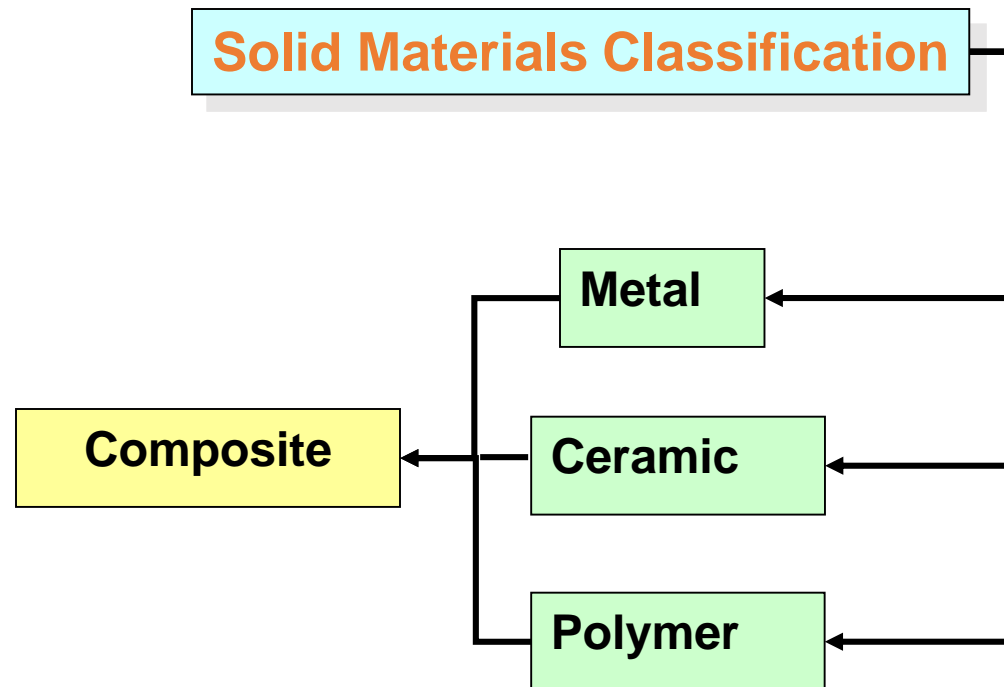
- “Tetrahedron” Interrelationship:
  - Car body → What is the right material to use?





# ■ 1.4 Class of Materials

# Class of Materials



## METALS:

- Inorganic materials (one or more metallic elements – alloy)
- Crystalline structure
- May contain a small amount of non-metallic elements
- Good thermal & electrical conductors (large numbers of non-localized electrons = electrons are not bound to particular atoms)
- Properties at room temperature (RT)
  - Strong and ductile (capable of large amounts of deformation without fracture)
  - Stiff
  - Good strength
  - Dense
  - Resistance to fracture



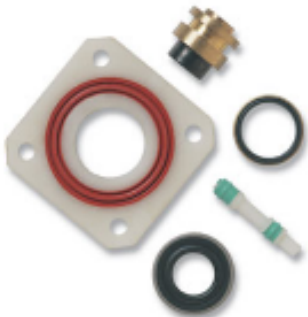
## ■ CERAMICS:

- Inorganic materials (metallic + non-metallic elements)
- Oxides, nitrides, carbides
- Crystalline, non-crystalline, or a mixture of both
- Properties at RT
  - High melting point
  - High chemical stabilities
  - High hardness
  - High temperature strength
  - Brittle (lack ductility)
  - Poor electrical and thermal conductor.



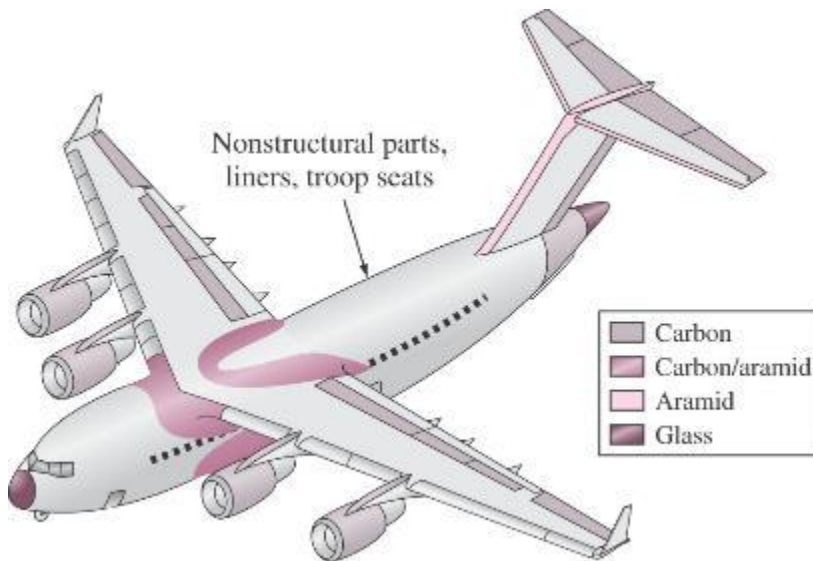
## ■ POLYMERS:

- Plastic and rubber materials
- Organic materials
- Long molecular chains/network containing C, H, and other nonmetallic elements (O, N, Si)
- Non-crystalline (mostly) or a mixture of both
- Properties at RT
  - Low density
  - Mechanically flexible
  - Poor electrical conductor



## ■ COMPOSITES:

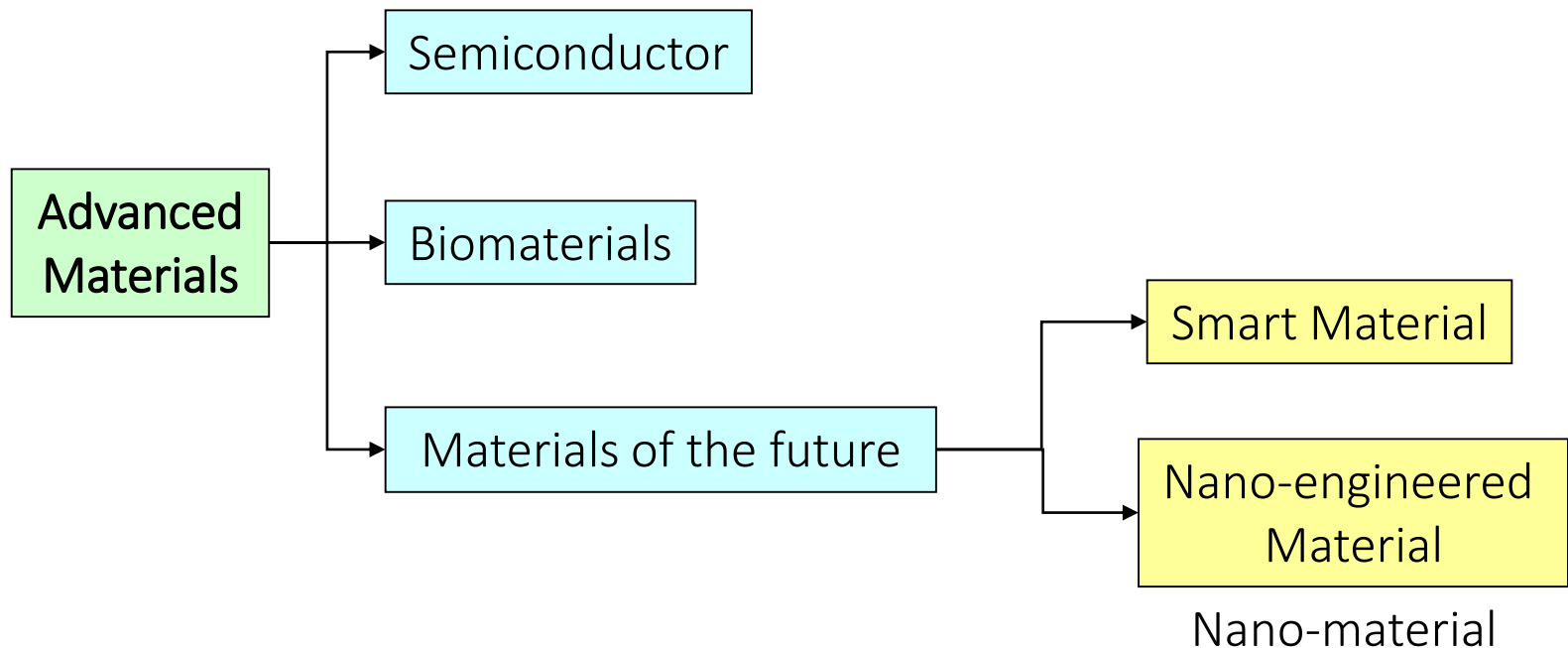
- Mixture of two or more types of materials
- A matrix phase + a reinforcing phase
- Designed to ensure a combination of the best properties of each component material.



■ Besides the above mentioned classification, we also have:

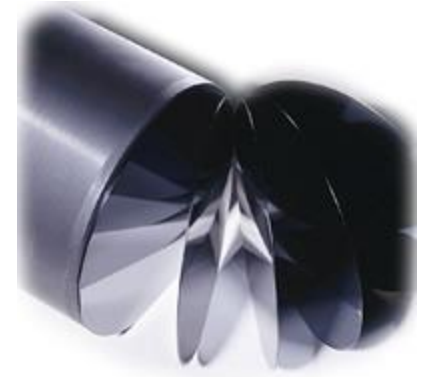
■ **ADVANCED MATERIALS:**

- Materials that are utilized in **high-tech application** (device/product that operates or functions using relatively intricate and sophisticated principles)
- DVD Players, Microprocessor, Liquid Crystal Display.



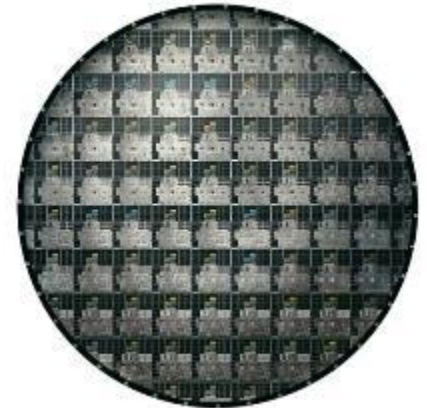
## ■ SEMICONDUCTORS:

- Electrical properties intermediate between conductors and insulators
- Electrical characteristics are extremely sensitive to the presence of minute concentration of impurity atoms, which concentrations may be controlled over very small spatial region
- Conductivity increasing with temperature



## ■ BIOMATERIALS:

- Components implanted into human body for replacement of diseased or damaged parts.
- Must not produce toxic substances
- Compatible with body tissues
- All above materials may be used as biomaterials





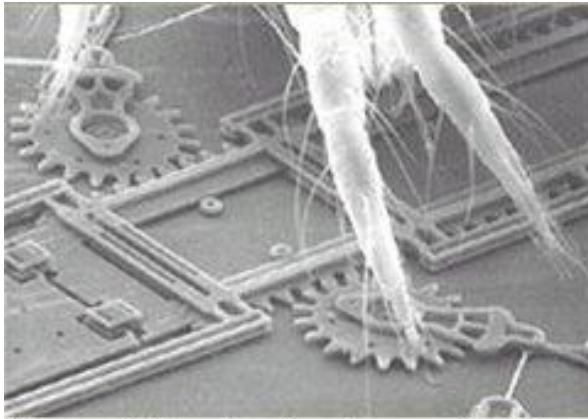
- **SMART MATERIALS:**

- Materials that are able to sense changes in their environments and then respond to these changes in predetermined manners

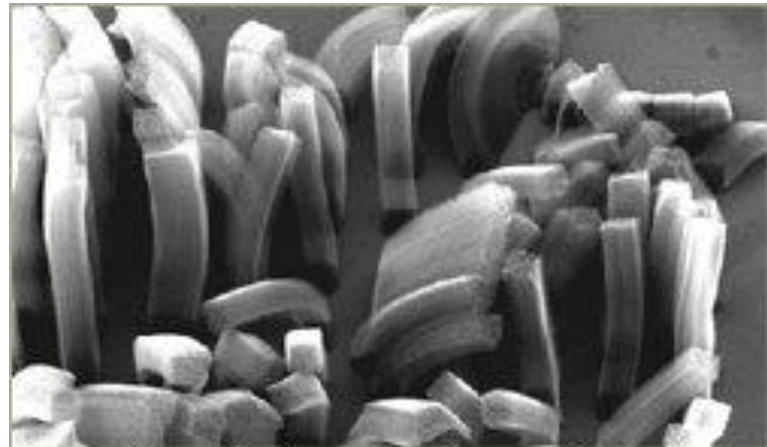
Devices made from Smart Materials

- Sensors (detects an input signal)

- **NANO-ENGINEERED MATERIALS:**
  - Dimension  $<100$  nm ( $\sim 500$  atom diameters)
  - Bottom-up technique
  - Top-down technique
  - Materials by design
  - Exp: carbon nanotube



Micromachining involves the creation of microscopic mechanical devices, like that shown here. The legs belong to a spider mite placed to demonstrate scale. Courtesy: Sandia National Laboratories.



Nanotubes, which are made in a flask by a chemical process. Courtesy: National Center of Competence in Research (Switzerland).

# QUIZ.....

- What is Materials science? What is material science tetrahedron?
- What are the major properties of Nickel base superalloys that make it suitable for aircraft turbine engines?



Answer: Excellent mechanical strength, creep resistance at high temperature, good surface stability, corrosion resistance