

BIOREACTOR ENGINEERING Chapter 5 Culture Kinetic Study of Modifying Batch Fermentation

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Culture Kinetic Study of Modifying Batch Fermentation by Chew Few Ne

Chapter Description

- Topic Outcome
 - Perform culture kinetic calculation on modifying batch bioreactor operation
- References
 - Doran, P.M. (2013) Bioprocess Engineering Principles. Elsevier.
 - Liu, S. (2013) Bioprocess Engineering: Kinetics,
 Biosystem, Sustainability and Reactor Design. Elsevier.
 - Rao, D.G. (2010) Introduction to Biochemical Engineering. McGraw Hill.

Topic Outline

• Fed-batch Operation



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- In fed-batch cultures, substrate is fed gradually into the culture until the maximum liquid fermenter volume is reached
- No removal of the culture until the end of the process
- The concentration of substrate fed into the culture can be controlled by changing the feed rate.





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ADVANTAGES:

- Fed-batch reactor can maintain LOW limiting substrate concentration. They are thus suited for producing product/cells when:
 - The substrate is inhibitory.
 - The oxygen uptake rate must be restricted/controlled.
 - The product/cell yields are highest at low substrate concentrations.
 - Product/cell formation is dependent on a specific nutrient composition.



ADVANTAGES:

- Fed-batch reactor can maintain LOW limiting substrate concentration. They are thus suited:
 - The extension of stationary phase.
 - Prolong microbe growth.
 - Feed does not need to contain all the nutrients needed to sustain growth.
 - No additional special piece of equipment required compared with the continuous fermentation mode of operation.
 - Can be operated in a various ways.
 - Allow the replacement of water loss (by evaporation) and decrease of the viscosity of the medium.



DISADVANTAGES:

- Higher operation cost.
- It requires previous analysis of the microorganisms.



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Material balance across the reactor:

Culture volume in – Culture volume out = Culture volume accumulation

 $F\rho_o - 0 = \frac{d(\rho V)}{dt}$

V_i= initial culture volume

F= constant flow rate of addition stream during fed-batch

p=density of the reactor contents

 ρ_0 = density of the feed stream

- Assume $\rho = \rho_0$ is constant.
- Rate of increase in culture volume is: $\frac{dV}{dt} = F$

Integrate: $V = V_i + Ft$

Material balance on the cells:

Cell maintenance and death are assumed to be small

Cell in – Cell out + Cell growth = Cell accumulation

 $FX_o - 0 + \mu XV = \frac{dVX}{dt}$

• $X_o = cell concentration in the feed = 0 &$

$$\frac{dV}{dt} = F$$

Please derive the formula to get $\rightarrow dX$

$$\frac{dX}{dt} = \mu X - DX$$

Material balance on the substrate:

Substrate in – Substrate out – Substrate consumed = Substrate accumulation

$$FS_{o} - 0 - \left(\frac{\mu X}{Y_{X/S}} + \frac{q_{p}X}{Y_{P/S}} + mX\right)V = \frac{dVS}{dt}$$

- $S_o = substrate concentration in the feed = 0 & <math>\frac{dV}{dt} = F$
- Assume:
 - $mX \ll \frac{\mu X}{Y_{X/S}}$ can be neglected.
 - If no product is formed

Please derive the formula to get \rightarrow

$$\frac{dS}{dt} = D(S_o - S) - \frac{\mu X}{Y_{X/S}}$$



- Fed-batch culture is an unsteady-state process. The values of D, V, X, S, and P inside the vessel may be changing with time.
- We assume fed-batch operation starts after the fermenter is operating as a batch process (cell concentration is high, substrate is exhausted).
- As substrate is fed, it is totally consumed, cell concentration remains high. So, $\frac{dX}{dt} = 0$

• The process reaches quasi-steady state (when nutrient consumption rate is nearly equal to nutrient feed rate).

• So,
$$\frac{dX}{dt} = 0$$
 and Monod equation: $\mu = \mu_{max} \frac{S}{K_s + S}$

• From mass balance on cells,

$$S \cong \frac{K_S D}{\mu_{max} - D}$$



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• The process reaches quasi-steady state (when nutrient consumption rate is nearly equal to nutrient feed rate).

• So,
$$\frac{dS}{dt} = 0$$

• From mass balance on substrate,

$$X \approx Y_{X/S} S_o$$



- The cell concentration in the vessel at any time t is: $X = \frac{X^{t}}{X}$
- The rate of change in cell concentration is:

 $\frac{dX}{dt} = \frac{d\left(\frac{X^{t}}{V}\right)}{dt} = \frac{V\left(\frac{dX^{t}}{dt}\right) - X^{t}\left(\frac{dV}{dt}\right)}{V^{2}}$

- At quasi-steady state, $\frac{dX}{dt} = 0$
- Rearrange: $\frac{dX^{t}}{dt} = \frac{X^{t}}{V} \frac{dV}{dt} = XF = Y_{X/S}S_{o}F$

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• Integrate: $X^{t} = X_{i}^{t} + Y_{X/S}S_{o}Ft$ OR $X^{t} = V_{i}X_{i} + XFt$

where

- X^t = mass of cells in the fermenter after t
- X^t_i = mass of cells in the fermenter at the initial of fed-batch feeding
- X_i = concentration of cells in the fermenter at the initial of fed-batch feeding
- t = fed-batch time after feeding commencement



Material balance on the product:

Product in – Product out + Product synthesis = Product accumulation

$$0 - 0 + q_p XV = \frac{dVP}{dt}$$

At quasi-steady state, $\frac{dP}{dt} = 0$

Please derive the formula to get \rightarrow

$$q_p X = DP$$



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• The product concentration in the vessel at any time t is:

$$P = \frac{P^t}{V}$$

• The rate of change in product concentration is:

$$\frac{dP}{dt} = \frac{d\left(\frac{P^{t}}{V}\right)}{dt} = \frac{V\left(\frac{dP^{t}}{dt}\right) - P^{t}\left(\frac{dV}{dt}\right)}{V^{2}}$$

- At quasi-steady state, $\frac{dP}{dt} = 0$
- Rearrange: $\frac{dP^t}{dt} = \frac{P^t}{V}\frac{dV}{dt} = PF$

• Combine
$$q_p X = DP$$
 $\frac{dP^t}{dt} = PF$

• We obtain:
$$\frac{dP^t}{dt} = q_p X^t$$

• From previous, $X^{t} = V_{i}X_{i} + XFt$

• Thus,
$$\frac{dP^t}{dt} = q_p (V_i X_i + XFt)$$

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• Integrate:
$$P^{t} = P_{i}^{t} + q_{p}t\left(V_{i}X_{i} + \frac{XFt}{2}\right)$$

$$P = P_i \frac{V_i}{V} + q_p t \left(\frac{V_i X_i}{V} + \frac{XDt}{2}\right)$$

where

- P^t = amount of product in the fermenter at t
- P_{i}^{t} = amount of product in the fermenter at the initial of fed-batch feeding
- P_i = concentration of product in the fermenter at the initial of fed-batch feeding
- t = fed-batch time after feeding commencement



- Exercise 1
- Exercise 2



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